

# Bismuth(III) nitrate pentahydrate Apollo Scientific

Part Number: IN1250 Version No: 1.1 Safety Data Sheet (Conforms to Annex II of REACH (1907/2006) - Regulation 2020/878) Chemwatch Hazard Alert Code: 2

Issue Date: **16/09/2022** Print Date: **31/07/2023** S.REACH.GBR.EN

# SECTION 1 Identification of the substance / mixture and of the company / undertaking

#### 1.1. Product Identifier

Product name	Bismuth(III) nitrate pentahydrate
Chemical Name	bismuth nitrate
Synonyms	Not Available
Proper shipping name	NITRATES, INORGANIC, N.O.S.
Chemical formula	BiN3O9-5H2O Bi .3 H-N-O3
Other means of identification	Not Available
CAS number	10035-06-0

#### 1.2. Relevant identified uses of the substance or mixture and uses advised against

Relevant identified uses	Not Available
Uses advised against	No specific uses advised against are identified.

### 1.3. Details of the manufacturer or supplier of the safety data sheet

Registered company name	Apollo Scientific		
Address	Whitefield Road, Bredbury SK62QR United Kingdom		
Telephone	01614060505		
Fax	0161 406 0506		
Website	http://www.apolloscientific.co.uk/		
Email	sales@apolloscientific.co.uk		

#### 1.4. Emergency telephone number

Association / Organisation	Not Available
Emergency telephone numbers	Not Available
Other emergency telephone numbers	Not Available

#### **SECTION 2 Hazards identification**

#### 2.1. Classification of the substance or mixture

Classification according to regulation (EC) No 1272/2008 [CLP] and amendments <sup>[1]</sup>

H335 - Specific Target Organ Toxicity - Single Exposure (Respiratory Tract Irritation) Category 3, H315 - Skin Corrosion/Irritation Category 2, H319 - Serious Eye Damage/Eye Irritation Category 2, H272 - Oxidizing Solids Category 2

Legend: 1. Classified by Chemwatch; 2. Classification drawn from Regulation (EU) No 1272/2008 - Annex VI

# 2.2. Label elements

Hazard pictogram(s)	
Signal word	Danger

#### Hazard statement(s)

H335	May cause respiratory irritation.
H315	Causes skin irritation.
H319	Causes serious eye irritation.
H272	May intensify fire; oxidiser.

#### Supplementary statement(s)

Not Applicable

#### Precautionary statement(s) Prevention

P210	Keep away from heat, hot surfaces, sparks, open flames and other ignition sources. No smoking.		
P271	Use only outdoors or in a well-ventilated area.		
P220	Keep away from clothing and other combustible materials.		
P261	Avoid breathing dust/fumes.		
P280	Wear protective gloves, protective clothing, eye protection and face protection.		
P264	Wash all exposed external body areas thoroughly after handling.		

#### Precautionary statement(s) Response

P370+P378	In case of fire: Use to extinguish.
P305+P351+P338	IF IN EYES: Rinse cautiously with water for several minutes. Remove contact lenses, if present and easy to do. Continue rinsing.
P312	Call a POISON CENTER/doctor/physician/first aider/if you feel unwell.
P337+P313	If eye irritation persists: Get medical advice/attention.
P302+P352	IF ON SKIN: Wash with plenty of water.
P304+P340	IF INHALED: Remove person to fresh air and keep comfortable for breathing.
P332+P313	If skin irritation occurs: Get medical advice/attention.
P362+P364	Take off contaminated clothing and wash it before reuse.

#### Precautionary statement(s) Storage

P405	Store locked up.
P403+P233	Store in a well-ventilated place. Keep container tightly closed.

#### Precautionary statement(s) Disposal

P501 Dispose of contents/container to authorised hazardous or special waste collection point in accordance with any local regulation.

#### 2.3. Other hazards

May produce discomfort of the respiratory system and skin\*.

REACH - Art.57-59: The mixture does not contain Substances of Very High Concern (SVHC) at the SDS print date.

## **SECTION 3 Composition / information on ingredients**

3.1.Substances

1. CAS No 2.EC No

SCL / M-Factor Nanoform Particle Characteristics

Continued...

3.Index No 4.REACH No			amendments		
Not Available	100	Bismuth(III) nitrate pentahydrate	Not Applicable	Not Applicable	Not Available

Legend: 1. Classified by Chemwatch; 2. Classification drawn from Regulation (EU) No 1272/2008 - Annex VI; 3. Classification drawn from C&L; \* EU IOELVs available; [e] Substance identified as having endocrine disrupting properties

#### 3.2.Mixtures

See 'Information on ingredients' in section 3.1

#### **SECTION 4 First aid measures**

### 4.1. Description of first aid measures

Eye Contact	<ul> <li>If this product comes in contact with the eyes:</li> <li>Wash out immediately with fresh running water.</li> <li>Ensure complete irrigation of the eye by keeping eyelids apart and away from eye and moving the eyelids by occasionally lifting the upper and lower lids.</li> <li>Seek medical attention without delay; if pain persists or recurs seek medical attention.</li> <li>Removal of contact lenses after an eye injury should only be undertaken by skilled personnel.</li> </ul>
Skin Contact	<ul> <li>If skin contact occurs:</li> <li>Immediately remove all contaminated clothing, including footwear.</li> <li>Flush skin and hair with running water (and soap if available).</li> <li>Seek medical attention in event of irritation.</li> </ul>
Inhalation	<ul> <li>If fumes, aerosols or combustion products are inhaled remove from contaminated area.</li> <li>Other measures are usually unnecessary.</li> </ul>
Ingestion	<ul> <li>Immediately give a glass of water.</li> <li>First aid is not generally required. If in doubt, contact a Poisons Information Centre or a doctor.</li> </ul>

#### 4.2 Most important symptoms and effects, both acute and delayed

See Section 11

#### 4.3. Indication of any immediate medical attention and special treatment needed

Treat symptomatically.

The toxicity of nitrates and nitrites result from their vasodilating properties and their propensity to form methaemoglobin.

- Most produce a peak effect within 30 minutes.
- Clinical signs of cyanosis appear before other symptoms because of the dark pigmentation of methaemoglobin.
- Initial attention should be directed towards improving oxygen delivery, with assisted ventilation, if necessary. Hyperbaric oxygen has not demonstrated conclusive benefits.
- ▶ Institute cardiac monitoring, especially in patients with coronary artery or pulmonary disease.
- + Hypotension should respond to Trendelenburg's position and intravenous fluids; otherwise dopamine may be needed.
- ▶ Naloxone, glucose and thiamine should be given if a multiple ingestion is suspected.
- Decontaminate using Ipecac Syrup for alert patients or lavage for obtunded patients who present within 2-4 hours of ingestion.
- Symptomatic patients with methaemoglobin levels over 30% should receive methylene blue.(Cyanosis alone, is not an indication for treatment). The usual dose is 1-2 mg/kg of a 1% solution (10 mg/ml) IV over 5 minutes; repeat, using the same dose if symptoms of hypoxia fail to subside within 1 hour.

[Ellenhorn and Barceloux: Medical Toxicology]

**BIOLOGICAL EXPOSURE INDEX - BEI** 

These represent the determinants observed in specime	ns collected from a healthy worker who has	been exposed at the Exposure Standard	(ES or TLV):
Determinant	Index	Sampling Time	Comments
1. Methaemoglobin in blood	1.5% of haemoglobin	During or end of shift	B,NS,SQ

B: Background levels occur in specimens collected from subjects **NOT** exposed

NS: Non-specific determinant; also observed after exposure to other materials

SQ: Semi-quantitative determinant - Interpretation may be ambiguous; should be used as a screening test or confirmatory test.

#### **SECTION 5 Firefighting measures**

#### 5.1. Extinguishing media

#### FOR SMALL FIRE:

▶ USE FLOODING QUANTITIES OF WATER.

**DO NOT** use dry chemical, CO2, foam or halogenated-type extinguishers.

FOR LARGE FIRE

Flood fire area with water from a protected position

#### 5.2. Special hazards arising from the substrate or mixture

Fire Incompatibility	Fire	Incompatibility	
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Avoid storage with reducing agents.
Avoid any contamination of this material as it is very reactive and any contamination is potentially hazardous

#### 5.3. Advice for firefighters

Fire Fighting	
Fire/Explosion Hazard	<ul> <li>Will not burn but increases intensity of fire.</li> <li>Heating may cause expansion or decomposition leading to violent rupture of containers.</li> <li>Heat affected containers remain hazardous.</li> <li>Contact with combustibles such as wood, paper, oil or finely divided metal may produce spontaneous combustion or violent decomposition.</li> <li>May emit irritating, poisonous or corrosive fumes.</li> </ul>

## **SECTION 6 Accidental release measures**

## 6.1. Personal precautions, protective equipment and emergency procedures

See section 8

#### 6.2. Environmental precautions

See section 12

## 6.3. Methods and material for containment and cleaning up

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Minor Spills	<ul> <li>Clean up all spills immediately.</li> <li>No smoking, naked lights, ignition sources.</li> <li>Avoid all contact with any organic matter including fuel, solvents, sawdust, paper or cloth and other incompatible materials, as ignition may result.</li> <li>Avoid breathing dust or vapours and all contact with skin and eyes.</li> <li>Control personal contact with the substance, by using protective equipment.</li> <li>Contain and absorb spill with dry sand, earth, inert material or vermiculite.</li> <li>DO NOT use sawdust as fire may result.</li> <li>Scoop up solid residues and seal in labelled drums for disposal.</li> <li>Neutralise/decontaminate area.</li> </ul>
Major Spills	<ul> <li>Clear area of personnel and move upwind.</li> <li>Alert Fire Brigade and tell them location and nature of hazard.</li> <li>May be violently or explosively reactive.</li> <li>Wear full body protective clothing with breathing apparatus.</li> <li>Prevent, by any means available, spillage from entering drains or water course.</li> <li>Consider evacuation (or protect in place).</li> <li>No smoking, flames or ignition sources.</li> <li>Increase ventilation.</li> <li>Contain spill with sand, earth or other clean, inert materials.</li> <li>NEVER use organic absorbents such as sawdust, paper, cloth; as fire may result.</li> <li>Avoid any contamination by organic matter.</li> <li>Use spark-free and explosion-proof equipment.</li> <li>Collect any recoverable product into labelled containers for possible recycling.</li> <li>DO NOT mix fresh with recovered material.</li> <li>Collect residues and seal in labelled drums for disposal.</li> <li>Wash area and prevent runoff into drains.</li> <li>Decontaminate equipment and launder all protective clothing before storage and re-use.</li> <li>If contamination of drains or waterways occurs advise emergency services.</li> </ul>

### 6.4. Reference to other sections

Personal Protective Equipment advice is contained in Section 8 of the SDS.

## **SECTION 7 Handling and storage**

#### 7.1. Precautions for safe handling

Safe handling

g For oxidisers, including peroxides.

	Avoid personal contact and inhalation of dust, mist or vapours.
	Provide adequate ventilation.
	Always wear protective equipment and wash off any spillage from clothing.
	Keep material away from light, heat, flammables or combustibles.
	Keep cool, dry and away from incompatible materials.
	<ul> <li>Avoid physical damage to containers.</li> <li>DO NOT repack or return unused portions to original containers. Withdraw only sufficient amounts for immediate use.</li> </ul>
	· Use only minimum quantity required.
	Avoid using solutions of peroxides in volatile solvents. Solvent evaporation should be controlled to avoid dangerous
	concentration of the peroxide.
	• Do NOT allow oxidisers to contact iron or compounds of iron, cobalt, or copper, metal oxide salts, acids or bases.
	• Do NOT use metal spatulas to handle oxidisers
	Do NOT use glass containers with screw cap lids or glass stoppers.
	· Store peroxides at the lowest possible temperature, consistent with their solubility and freezing point.
	· CAUTION: Do NOT store liquids or solutions of peroxides at a temperature below that at which the oxidiser freezes or
	precipitates. Peroxides, in particular, in this form are extremely shock and heat-sensitive. Refrigerated storage of peroxides must
	ONLY be in explosion-proof units.
	· The hazards and consequences of fires and explosions during synthesis and use of oxidisers is widely recognised;
	spontaneous or induced decomposition may culminate in a variety of ways, ranging from moderate gassing to spontaneous
	ignition or explosion. The heat released from spontaneous decomposition of an energy-rich compound causes a rise in the
	surrounding temperature; the temperature will rise until thermal balance is established or until the material heats to
	decomposition,
	• The most effective means for minimising the consequences of an accident is to limit quantities to a practical minimum. Even
	gram-scale explosions can be serious. Once ignited the burning of peroxides cannot be controlled and the area should be
	evacuated.
	Unless there is compelling reason to do otherwise, peroxide concentration should be limited to 10% (or less with vigorous reactants). Beravide concentration is rarely as high as 1% in the reaction mixture of polymerication or other free radical reaction
	reactants). Peroxide concentration is rarely as high as 1% in the reaction mixture of polymerisation or other free-radical reaction Oxidisors should be added clowly and cautiously to the reaction medium. This should be completed prior to beating and with
	<ul> <li>Oxidisers should be added slowly and cautiously to the reaction medium. This should be completed prior to heating and with good agitation.</li> </ul>
	Addition oxidisers to the hot monomer is extremely dangerous. A violent reaction (e.g., fire or explosion) can result from
	inadvertent mixing of promoters (frequently used with peroxides in polymerisation systems) with full-strength oxidisers
	· Organic peroxides are very sensitive to contamination (especially heavy-metal compounds, metal oxide salts, alkaline materia
	including amines, strong acids, and many varieties of dust and dirt). This can initiate rapid, uncontrolled decomposition of
	peroxides and possible generation of intense heat, fire or explosion The consequences of accidental contamination from
	returning withdrawn material to the storage container can be disastrous.
	When handling NEVER smoke, eat or drink.
	· Always wash hands with soap and water after handling.
	Use only good occupational work practice.
	Observe manufacturer's storage and handling recommendations contained within this MSDS.
Fire and explosion protection	See section 5
	Store in original containers.
	<ul> <li>Keep containers securely sealed as supplied.</li> </ul>
	Store in a cool, well ventilated area.
	► Keep dry.
	Store under cover and away from sunlight.
	Store away from flammable or combustible materials, debris and waste. Contact may cause fire or violent reaction.
Other information	Store away from incompatible materials and foodstuff containers.
	DO NOT stack on wooden floors or pallets.
	Protect containers from physical damage.
	Check regularly for leaks.
	Observe manufacturer's storage and handling recommendations contained within this SDS.
	In addition, Goods of Class 5.1, packing group III should be stored in packages and be separated from buildings, tanks, and
	compounds containing other dangerous goods in tanks, and from property boundaries by a distance of at least 5 metres.

	DO NOT repack. Use containers supplied by manufacturer only.
	For low viscosity materials
	Drums and jerricans must be of the non-removable head type.
	Where a can is to be used as an inner package, the can must have a screwed enclosure.
	For materials with a viscosity of at least 2680 cSt. (23 deg. C) and solids:
	Removable head packaging and
	cans with friction closures may be used.
Suitable container	-
	Where combination packages are used, and the inner packages are of glass, there must be sufficient inert cushioning material in
	contact with inner and outer packages *.
	-
	In addition, where inner packagings are glass and contain liquids of packing group I and II there must be sufficient inert
	absorbent to absorb any spillage *.

	* unless the outer packaging is a close fitting moulded plastic box and the substances are not incompatible with the plastic.
Storage incompatibility	<ul> <li>Inorganic peroxy compounds are potent oxidisers that pose fire or explosive hazards when in contact with ordinary combustible materials.</li> <li>Inorganic peroxides react with organic compounds to generate organic peroxide and hydroperoxide products that react violently with reducing agents.</li> <li>Inorganic oxidising agents can react with reducing agents to generate heat and products that may be gaseous (causing pressurization of closed containers). The products may themselves be capable of further reactions (such as combustion in the air).</li> <li>Organic compounds in general have some reducing power and can in principle react with compounds in this class. Actual reactivity varies greatly with the identity of the organic compound.</li> <li>Inorganic oxidising agents can react violently with active metals, cyanides, esters, and thiocyanates.</li> <li>Peroxides, in contact with inorganic cobalt and copper compounds, iron and iron compounds, acene, metal oxide salts and acids and bases can react with toxidizing agents to generate heat and products that may be flammable, combustible, or otherwise reactive. Their reactions with oxidizing agents may be violent.</li> <li>Incigentis involving interaction of active oxidants and reducing agents, either by design or accident, are usually very energetic and examples of so-called redox reactions.</li> <li>Segregate from heavy metals, phosphides, sodium acetate, lead nitrate, tartrates, trichloroethylene,</li> <li>Avoid shock and heat.</li> <li>Mixtures of metal nitrates with phosphorous, tin(II) choride and other reducing agents may react explosively.</li> <li>Mixtures of metal nitrates and organic materials are potentially dangerous, especially if acidic materials or heavy metals ar present.</li> <li>Fibrous organic material, jute, wood and similar cellulosic material can become highly combustible by nitrate impregnation</li> <li>Metal nitrates are incompatible with cyanides, thoicyanates, isochino-cyanates, and hypophosphites. Avoid reaction with the follow</li></ul>
Hazard categories in accordance with Regulation (EC) No 1272/2008	Not Available
Qualifying quantity (tonnes) of dangerous substances as referred to in Article 3(10) for the application of	Not Available

# 7.3. Specific end use(s)

See section 1.2

# **SECTION 8 Exposure controls / personal protection**

# 8.1. Control parameters

Ingredient	DNELs Exposure Pattern Worker	PNECs Compartment
Not Available	Dermal 30 mg/kg bw/day (Systemic, Chronic) Inhalation 21.1 mg/m <sup>3</sup> (Systemic, Chronic) Dermal 15 mg/kg bw/day (Systemic, Chronic) * Inhalation 5.22 mg/m <sup>3</sup> (Systemic, Chronic) * Oral 1.5 mg/kg bw/day (Systemic, Chronic) *	0.137 mg/L (Water (Fresh)) 0.014 mg/L (Water - Intermittent release) 1.37 mg/L (Water (Marine)) 14176.5 mg/kg sediment dw (Sediment (Fresh Water)) 1417.7 mg/kg sediment dw (Sediment (Marine)) 120.3 mg/kg soil dw (Soil) 17.5 mg/L (STP) 33.3 mg/kg food (Oral)

\* Values for General Population

## **Occupational Exposure Limits (OEL)**

# INGREDIENT DATA

Source Ingredient	Material name	TWA	STEL	Peak	Notes
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Source	Ingredient	Material name	TWA	STEL	Peak	Notes
Not Available						

Not Applicable

### Emergency Limits

Ingredient	TEEL-1	TEEL-2		TEEL-3
Bismuth(III) nitrate pentahydrate	Not Available	Not Available		Not Available
Ingredient	Original IDLH		Revised IDLH	
Bismuth(III) nitrate pentahydrate	Not Available		Not Available	

## 8.2. Exposure controls

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	Engineering controls are used to remove a hazard or place a engineering controls can be highly effective in protecting wor provide this high level of protection. The basic types of engineering controls are: Process controls which involve changing the way a job activi Enclosure and/or isolation of emission source which keeps a that strategically "adds" and "removes" air in the work enviro designed properly. The design of a ventilation system must re Employers may need to use multiple types of controls to pre-	kers and will typically be independent of wor ty or process is done to reduce the risk. selected hazard "physically" away from the nment. Ventilation can remove or dilute an a natch the particular process and chemical or	ker interactions to worker and ventilation ir contaminant if		
	Local exhaust ventilation usually required. If risk of overexposure exists, wear approved respirator. Correct fit is essential to obtain adequate protection. Supplied-air type respirator may be required in special circumstances. Correct fit is essential to ensure adequate protection. An approved self contained breathing apparatus (SCBA) may be required in some situations. Provide adequate ventilation in warehouse or closed storage area. Air contaminants generated in the workplace possess varying "escape" velocities which, in turn, determine the "capture velocities" of fresh circulating air required to effectively remove the contaminant.				
	Type of Contaminant:		Air Speed:		
	solvent, vapours, degreasing etc., evaporating from tank (i	0.25-0.5 m/s (50-100 f/min.)			
1. Appropriate ering controls	aerosols, fumes from pouring operations, intermittent container filling, low speed conveyer transfers, welding, spray drift, plating acid fumes, pickling (released at low velocity into zone of active generation)		0.5-1 m/s (100-200 f/min.)		
	direct spray, spray painting in shallow booths, drum filling, conveyer loading, crusher dusts, gas discharge (active generation into zone of rapid air motion)		1-2.5 m/s (200-500 f/min.)		
	grinding, abrasive blasting, tumbling, high speed wheel generated dusts (released at high initial velocity into zone of very high rapid air motion).		2.5-10 m/s (500-2000 f/min.)		
	Within each range the appropriate value depends on:				
	Lower end of the range	Upper end of the range			
	1: Room air currents minimal or favourable to capture	1: Disturbing room air currents			
	2: Contaminants of low toxicity or of nuisance value only.	2: Contaminants of high toxicity			
	3: Intermittent, low production.	3: High production, heavy use			
	4: Large hood or large air mass in motion				
	Simple theory shows that air velocity falls rapidly with distance generally decreases with the square of distance from the ext extraction point should be adjusted, accordingly, after referent extraction fan, for example, should be a minimum of 1-2 m/s	raction point (in simple cases). Therefore the nee to distance from the contaminating source	e air speed at the e. The air velocity at nerated in a tank 2		

8.2.2. Individual protection measures, such as personal protective equipment installed or used.

Eye and face protection



- Chemical goggles. [AS/NZS 1337.1, EN166 or national equivalent]
- ► Full face shield may be required for supplementary but never for primary protection of eyes.
- Contact lenses may pose a special hazard; soft contact lenses may absorb and concentrate irritants. A written policy document, describing the wearing of lenses or restrictions on use, should be created for each workplace or task. This should

	include a review of lens absorption and adsorption for the class of chemicals in use and an account of injury experience. Medical and first-aid personnel should be trained in their removal and suitable equipment should be readily available. In the event of chemical exposure, begin eye irrigation immediately and remove contact lens as soon as practicable. Lens should be removed at the first signs of eye redness or irritation - lens should be removed in a clean environment only after workers have washed hands thoroughly. [CDC NIOSH Current Intelligence Bulletin 59].
Skin protection	See Hand protection below
Hands/feet protection	<ul> <li>Wear chemical protective gloves, e.g. PVC.</li> <li>Wear safety footwear or safety gumboots, e.g. Rubber</li> <li>The selection of suitable gloves does not only depend on the material, but also on further marks of quality which vary from manufacturer to manufacturer. Where the chemical is a preparation of several substances, the resistance of the glove material can not be calculated in advance and has therefore to be checked prior to the application.</li> <li>The exact break through time for substances has to be obtained from the manufacturer of the protective gloves and has to be observed when making a final choice.</li> <li>Personal hygiene is a key element of effective hand care. Gloves must only be worn on clean hands. After using gloves, hands should be washed and dired thoroughly. Application of a non-perfumed moisturiser is recommended.</li> <li>Suitability and durability of glove type is dependent on usage. Important factors in the selection of gloves include:         <ul> <li>Frequency and duration of contact,</li> <li>elemeical resistance of glove material,</li> <li>glove thickness and</li> <li>dexterity</li> </ul> </li> <li>Select gloves tested to a relevant standard (e.g. Europe EN 374, US F739, AS/NZS 2161.1 or national equivalent).</li> <li>When prolonged or frequently repeated contact may occur, a glove with a protection class of 5 or higher (breakthrough time greater than 240 minutes according to EN 374, AS/NZS 2161.1.0.1 or national equivalent) is recommended.</li> <li>Wome glove polymer types are less affected by movement and this should be taken into account when considering gloves for long-term use.</li> <li>Contaminated gloves should be replaced.</li> <li>As defined in ASTM F.739-96 in any application, gloves are rated as:</li> <li>Excellent when breakthrough time &lt; 20 min</li> <li>Fair when breakthrough time &lt; 20 min</li> <li>Fair when breakthrough time &lt; 20 min</li></ul>
Body protection	See Other protection below
Other protection	<ul> <li>Overalls.</li> <li>PVC Apron.</li> <li>PVC protective suit may be required if exposure severe.</li> <li>Eyewash unit.</li> <li>Ensure there is ready access to a safety shower.</li> <li>Some plastic personal protective equipment (PPE) (e.g. gloves, aprons, overshoes) are not recommended as they may produce static electricity.</li> <li>For large scale or continuous use wear tight-weave non-static clothing (no metallic fasteners, cuffs or pockets).</li> <li>Non sparking safety or conductive footwear should be considered. Conductive footwear describes a boot or shoe with a sole made from a conductive compound chemically bound to the bottom components, for permanent control to electrically ground the foot an shall dissipate static electricity from the body to reduce the possibility of ignition of volatile compounds. Electrical resistance must range between 0 to 500,000 ohms. Conductive shoes should be stored in lockers close to the room in which they are worn. Personnel who have been issued conductive footwear should not wear them from their place of work to their homes and return.</li> </ul>

# **Respiratory protection**

Type -P Filter of sufficient capacity. (AS/NZS 1716 & 1715, EN 143:2000 & 149:2001, ANSI Z88 or national equivalent)

Required Minimum Protection Factor	Half-Face Respirator	Full-Face Respirator	Powered Air Respirator
	nun ruoc neophator	i un i doc nospirator	r owered An Respirator

up to 10 x ES	P1 Air-line*	-	PAPR-P1 -
up to 50 x ES	Air-line**	P2	PAPR-P2
up to 100 x ES	-	P3	-
		Air-line*	-
100+ x ES	-	Air-line**	PAPR-P3

\* - Negative pressure demand \*\* - Continuous flow

A(All classes) = Organic vapours, B AUS or B1 = Acid gasses, B2 = Acid gas or hydrogen cyanide(HCN), B3 = Acid gas or hydrogen cyanide(HCN), E = Sulfur dioxide(SO2), G = Agricultural chemicals, K = Ammonia(NH3), Hg = Mercury, NO = Oxides of nitrogen, MB = Methyl bromide, AX = Low boiling point organic compounds(below 65 degC)

· Respirators may be necessary when engineering and administrative controls do not adequately prevent exposures.

• The decision to use respiratory protection should be based on professional judgment that takes into account toxicity information, exposure measurement data, and frequency and likelihood of the worker's exposure - ensure users are not subject to high thermal loads which may result in heat stress or distress due to personal protective equipment (powered, positive flow, full face apparatus may be an option).

• Published occupational exposure limits, where they exist, will assist in determining the adequacy of the selected respiratory protection. These may be government mandated or vendor recommended.

• Certified respirators will be useful for protecting workers from inhalation of particulates when properly selected and fit tested as part of a complete respiratory protection program.

• Where protection from nuisance levels of dusts are desired, use type N95 (US) or type P1 (EN143) dust masks. Use respirators and components tested and approved under appropriate government standards such as NIOSH (US) or CEN (EU)

· Use approved positive flow mask if significant quantities of dust becomes airborne.

· Try to avoid creating dust conditions.

#### 8.2.3. Environmental exposure controls

See section 12

#### **SECTION 9** Physical and chemical properties

# 9.1. Information on basic physical and chemical properties

Appearance	Not Available		
Physical state	Solid	Relative density (Water = 1)	Not Available
Odour	Not Available	Partition coefficient n-octanol / water	Not Available
Odour threshold	Not Available	Auto-ignition temperature (°C)	Not Available
pH (as supplied)	Not Available	Decomposition temperature (°C)	Not Available
Melting point / freezing point (°C)	30	Viscosity (cSt)	Not Available
Initial boiling point and boiling range (°C)	75-80	Molecular weight (g/mol)	Not Available
Flash point (°C)	Not Available	Taste	Not Available
Evaporation rate	Not Available	Explosive properties	Not Available
Flammability	Not Available	Oxidising properties	Not Available
Upper Explosive Limit (%)	Not Available	Surface Tension (dyn/cm or mN/m)	Not Applicable
Lower Explosive Limit (%)	Not Available	Volatile Component (%vol)	Not Available
Vapour pressure (kPa)	Not Available	Gas group	Not Available
Solubility in water	Not Available	pH as a solution (1%)	Not Available
Vapour density (Air = 1)	2.83	VOC g/L	Not Available
Nanoform Solubility	Not Available	Nanoform Particle Characteristics	Not Available
Particle Size	Not Available		

# **SECTION 10 Stability and reactivity**

10.1.Reactivity	See section 7.2	
10.2. Chemical stability	<ul> <li>Unstable in the presence of incompatible materials.</li> <li>Product is considered stable under normal handling conditions.</li> <li>Prolonged exposure to heat.</li> <li>Hazardous polymerisation will not occur.</li> </ul>	
10.3. Possibility of hazardous reactions See section 7.2		
10.4. Conditions to avoid	See section 7.2	
10.5. Incompatible materials	See section 7.2	
10.6. Hazardous decomposition products	See section 5.3	

## **SECTION 11 Toxicological information**

## 11.1. Information on hazard classes as defined in Regulation (EC) No 1272/2008 Information on toxicological effects

Inhaled	There is some evidence to suggest that the material can cause respiratory irritation in some persons. The body's response to such irritation can cause further lung damage. Persons with impaired respiratory function, airway diseases and conditions such as emphysema or chronic bronchitis, may incur further disability if excessive concentrations of particulate are inhaled. If prior damage to the circulatory or nervous systems has occurred or if kidney damage has been sustained, proper screenings should be conducted on individuals who may be exposed to further risk if handling and use of the material result in excessive exposures.		
Ingestion	relatively harmless, but where bacteria are present and the envir converted to nitrite. The main sites of this reaction are the mouth infection) may also be of some toxicological importance. Adults have tolerated large doses of sodium nitrate and ammoniu in some cases repeated for several days for medical or experime amount of methaemoglobin in the blood, diarrhea and vomiting). associated with doses greater than 10 grams of nitrate ion. Dose 150 milligrams of nitrate ion per kilogram body weight) have been impairing delivery of oxygen to the tissues. This is the main acute The half-life in the body for an oral dose of nitrate is approximate The material has <b>NOT</b> been classified by EC Directives or other of the lack of corroborating animal or human evidence. The substance and/or its metabolites may bind to haemoglobin in "methaemoglobinemia", is a form of oxygen starvation (anoxia). Symptoms include cyanosis (a bluish discolouration skin and mu be evident until several hours after exposure. At about 15% concentration of blood methaemoglobin there is ob may be absent although euphoria, flushed face and headache ar little disability occurs other than that produced on physical exertion lightheadedness, increasingly severe headache, ataxia, rapid sha	he main concern with exposure to inorganic nitrate is its biological reduction to the reactive and toxic nitrite. Nitrate itself is elatively harmless, but where bacteria are present and the environment is anaerobic (lacking in oxygen), nitrate can be onverted to nitrite. The main sites of this reaction are the mouth and stomach, but nitrite formation in the bladder (urinary offection) may also be of some toxicological importance. dults have tolerated large doses of sodium nitrate and ammonium nitrate (> 100 milligrams of nitrate per kilogram body weight), on some cases repeated for several days for medical or experimental purposes, with only minor effects in some subjects (slight mount of methaemoglobin in the blood, diarrhea and vomiting). Death and severe effects of swallowing nitrate are generally ssociated with doses greater than 10 grams of nitrate ion. Doses of between 2 and 9 grams of nitrate ion (equivalent to 33 to 50 milligrams of nitrate ion per kilogram body weight) have been reported to cause methaemoglobin to be present in the blood, npairing delivery of oxygen to the tissues. This is the main acute toxic effect of nitrate and nitrite poisoning. he half-life in the body for an oral dose of nitrate is approximately 5 hours. Nitrate does not accumulate in the body. he material has <b>NOT</b> been classified by EC Directives or other classification systems as "harmful by ingestion". This is because f the lack of corroborating animal or human evidence. he substance and/or its metabolites may bind to haemoglobin inhibiting normal uptake of oxygen. This condition, known as nethaemoglobinemia", is a form of oxygen starvation (anoxia). ymptoms include cyanosis (a bluish discolouration skin and mucous membranes) and breathing difficulties. Symptoms may not	
Skin Contact	<ul> <li>There is some evidence to suggest that this material can cause inflammation of the skin on contact in some persons.</li> <li>This material can cause eye irritation and damage in some persons.</li> <li>Long-term exposure to the product is not thought to produce chronic effects adverse to the health (as classified by EC Directives).</li> </ul>		
Eye			
Chronic			
	ΤΟΧΙCITY	IRRITATION	
Bismuth(III) nitrate	Oral (Mouse) LD50; 3710 mg/kg <sup>[2]</sup>	Eye: adverse effect observed (irritating) <sup>[1]</sup>	
pentahydrate		Skin: no adverse effect observed (not irritating) <sup>[1]</sup>	

Legend: 1. Value obtained from Europe ECHA Registered Substances - Acute toxicity 2. Value obtained from manufacturer's SDS. Unless otherwise specified data extracted from RTECS - Register of Toxic Effect of chemical Substances

Acute Toxicity	×	Carcinogenicity	×
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Skin Irritation/Corrosion	Reproductivity	×
Serious Eye Damage/Irritation	✓ STOT - Single Exposure	<b>v</b>
Respiratory or Skin sensitisation	X STOT - Repeated Exposure	×
Mutagenicity	× Aspiration Hazard	×

Legend:

Data either not available or does not fill the criteria for classification
 Data available to make classification

# 11.2 Information on other hazards

## 11.2.1. Endocrine disrupting properties

No evidence of endocrine disrupting properties were found in the current literature.

### 11.2.2. Other information

See Section 11.1

## **SECTION 12 Ecological information**

#### 12.1. Toxicity

	Endpoint	Test Duration (hr)	Species	Value	Source
	EC50	72h	Algae or other aquatic plants	>1.26mg/l	2
Bismuth(III) nitrate	EC50	48h	Crustacea	>1.26mg/l	2
pentahydrate	ErC50	72h	Algae or other aquatic plants	>1.26mg/l	2
	LC50	96h	Fish	>100mg/l	2
	NOEC(ECx)	72h	Algae or other aquatic plants	1mg/l	2
Legend:	Extracted from 1. IUCLID Toxicity Data 2. Europe ECHA Registered Substances - Ecotoxicological Information - Aquatic			tic Toxicit	
4. US EPA, Ecotox database - Aquatic Toxicity Data 5. ECETOC Aquatic Hazard Assessment Data 6. NITE (Japan) - Bioconcentration Data 7. METI (Japan) - Bioconcentration Data 8. Vendor Data					

## 12.2. Persistence and degradability

Ingredient	Persistence: Water/Soil	Persistence: Air	
No Data available for all ingredients		No Data available for all ingredients	

#### 12.3. Bioaccumulative potential

Ingredient	Bioaccumulation
	No Data available for all ingredients

# 12.4. Mobility in soil

Ingredient	Mobility
	No Data available for all ingredients

# 12.5. Results of PBT and vPvB assessment

	Р	В	т
Relevant available data	Not Available	Not Available	Not Available
PBT	×	×	×
vPvB	×	×	×
PBT Criteria fulfilled?			No
vPvB			No

### 12.6. Endocrine disrupting properties

No evidence of endocrine disrupting properties were found in the current literature.

#### 12.7. Other adverse effects

No evidence of ozone depleting properties were found in the current literature.

# **SECTION 13 Disposal considerations**

### 13.1. Waste treatment methods

Product / Packaging disposal	<ul> <li>Recycle wherever possible or consult manufacturer for recycling options.</li> <li>Consult State Land Waste Management Authority for disposal.</li> <li>Bury residue in an authorised landfill.</li> <li>Recycle containers if possible, or dispose of in an authorised landfill.</li> </ul>
Waste treatment options	Not Available
Sewage disposal options	Not Available

## **SECTION 14 Transport information**

# Labels Required

	51
Marine Pollutant	NO
HAZCHEM	1Y

# Land transport (ADR-RID)

14.1. UN number or ID number	1477		
14.2. UN proper shipping name	NITRATES, INOR	GANIC, N.O.S.	
14.3. Transport hazard	Class	5.1	
class(es)	Subsidiary risk	Not Applicab	le
14.4. Packing group	ш		
14.5. Environmental hazard	Not Applicable		
	Hazard identifica	ation (Kemler)	50
14.6. Special precautions for user	Classification co	de	O2
	Hazard Label		5.1
	Special provision	าร	511
	Limited quantity		5 kg
	Tunnel Restriction	on Code	3 (E)

# Air transport (ICAO-IATA / DGR)

14.1. UN number	1477				
14.2. UN proper shipping name	Nitrates, inorganic, n.o.s.				
	ICAO/IATA Class	5.1			
14.3. Transport hazard class(es)	ICAO / IATA Subrisk	Not Applicable			
01000(00)	ERG Code	5L			
14.4. Packing group	Ш				
14.5. Environmental hazard	Not Applicable				
14.6. Special precautions	Special provisions		A3 A803		
for user	Cargo Only Packing Ir	nstructions	563		

Continued...

Cargo Only Maximum Qty / Pack	100 kg
Passenger and Cargo Packing Instructions	559
Passenger and Cargo Maximum Qty / Pack	25 kg
Passenger and Cargo Limited Quantity Packing Instructions	Y546
Passenger and Cargo Limited Maximum Qty / Pack	10 kg

#### Sea transport (IMDG-Code / GGVSee)

14.1. UN number	1477	
14.2. UN proper shipping name	NITRATES, INORGANIC, N.O.S.	
14.3. Transport hazard	IMDG Class	5.1
class(es)	IMDG Subrisk	Not Applicable
14.4. Packing group	Ш	
14.5. Environmental hazard	Not Applicable	
	EMS Number	F-A, S-Q
14.6. Special precautions for user	Special provisions	3 223
	Limited Quantities	5 kg

# Inland waterways transport (ADN)

14.1. UN number	1477	
14.2. UN proper shipping name	NITRATES, INORGANI	C, N.O.S.
14.3. Transport hazard class(es)	5.1 Not Applicable	
14.4. Packing group	ш	
14.5. Environmental hazard	Not Applicable	
14.6. Special precautions for user	Classification code	O2
	Special provisions	511
	Limited quantity	5 kg
	Equipment required	PP
	Fire cones number	0

## 14.7. Maritime transport in bulk according to IMO instruments

#### 14.7.1. Transport in bulk according to Annex II of MARPOL and the IBC code

Not Applicable

# 14.7.2. Transport in bulk in accordance with MARPOL Annex V and the IMSBC Code

	Product name	Group			
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# 14.7.3. Transport in bulk in accordance with the IGC Code

Product name	Ship Type
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# **SECTION 15 Regulatory information**

# 15.1. Safety, health and environmental regulations / legislation specific for the substance or mixture

This safety data sheet is in compliance with the following EU legislation and its adaptations - as far as applicable - : Directives 98/24/EC, - 92/85/EEC, - 94/33/EC, - 2008/98/EC, - 2010/75/EU; Commission Regulation (EU) 2020/878; Regulation (EC) No 1272/2008 as updated through ATPs.

# Information according to 2012/18/EU (Seveso III):

Seveso Category	Not Available

#### 15.2. Chemical safety assessment

For further information please look at the Chemical Safety Assessment and Exposure Scenarios prepared by your Supply Chain if available.

#### ECHA SUMMARY

Not Applicable

### **National Inventory Status**

National Inventory	Status
Australia - AIIC / Australia Non-Industrial Use	Not Available
Canada - DSL	Not Available
Canada - NDSL	Not Available
China - IECSC	Not Available
Europe - EINEC / ELINCS / NLP	Not Available
Japan - ENCS	Not Available
Korea - KECI	Not Available
New Zealand - NZIoC	Not Available
Philippines - PICCS	Not Available
USA - TSCA	Not Available
Taiwan - TCSI	Not Available
Mexico - INSQ	Not Available
Vietnam - NCI	Not Available
Russia - FBEPH	Not Available
Legend:	Yes = All CAS declared ingredients are on the inventory No = One or more of the CAS listed ingredients are not on the inventory. These ingredients may be exempt or will require registration.

### **SECTION 16 Other information**

Revision Date	16/09/2022
Initial Date	16/09/2022

#### Full text Risk and Hazard codes

#### Other information

Classification of the preparation and its individual components has drawn on official and authoritative sources as well as independent review by the Chemwatch Classification committee using available literature references.

The SDS is a Hazard Communication tool and should be used to assist in the Risk Assessment. Many factors determine whether the reported Hazards are Risks in the workplace or other settings. Risks may be determined by reference to Exposures Scenarios. Scale of use, frequency of use and current or available engineering controls must be considered.

For detailed advice on Personal Protective Equipment, refer to the following EU CEN Standards:

EN 166 Personal eye-protection

EN 340 Protective clothing

- EN 374 Protective gloves against chemicals and micro-organisms
- EN 13832 Footwear protecting against chemicals

EN 133 Respiratory protective devices

#### **Definitions and abbreviations**

PC - TWA: Permissible Concentration-Time Weighted Average PC - STEL: Permissible Concentration-Short Term Exposure Limit IARC: International Agency for Research on Cancer ACGIH: American Conference of Governmental Industrial Hygienists STEL: Short Term Exposure Limit TEEL: Temporary Emergency Exposure Limit. IDLH: Immediately Dangerous to Life or Health Concentrations ES: Exposure Standard **OSF: Odour Safety Factor** 

NOAEL :No Observed Adverse Effect Level

LOAEL: Lowest Observed Adverse Effect Level TLV: Threshold Limit Value LOD: Limit Of Detection OTV: Odour Threshold Value **BCF: BioConcentration Factors BEI: Biological Exposure Index** AIIC: Australian Inventory of Industrial Chemicals DSL: Domestic Substances List NDSL: Non-Domestic Substances List IECSC: Inventory of Existing Chemical Substance in China EINECS: European INventory of Existing Commercial chemical Substances ELINCS: European List of Notified Chemical Substances NLP: No-Longer Polymers ENCS: Existing and New Chemical Substances Inventory KECI: Korea Existing Chemicals Inventory NZIoC: New Zealand Inventory of Chemicals PICCS: Philippine Inventory of Chemicals and Chemical Substances TSCA: Toxic Substances Control Act TCSI: Taiwan Chemical Substance Inventory INSQ: Inventario Nacional de Sustancias Químicas NCI: National Chemical Inventory FBEPH: Russian Register of Potentially Hazardous Chemical and Biological Substances

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end of SDS